

Nitric oxide inhibitory activity of hydrogenated synthetic analogues of furanonaphthoquinones isolated from *Tabebuia* spp.

Efecto inhibidor de los análogos sintéticos hidrogenados de las furanoaftoquinonas aisladas del género *Tabebuia* spp. sobre la producción de óxido nítrico

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ABSTRACT

Objective: to describe the synthesis of analogues of furanonaphthoquinones isolated from *Tabebuia* genus and their inhibitory effect on nitric oxide production.

Methods: a series of six derivatives were prepared through cycloaddition reactions and the products characterized by spectroscopy methods. The biological activity was evaluated measuring their effect on the pro-inflammatory mediator production in macrophages RAW 264.7 induced with lipopolysaccharides. To prevent compounds from interfering with cellular viability, their cytotoxic effect was determined using methyl tetrazolium assay. Additionally, scavenging effect was *in vitro* measured.

Results: FNQ1, FNQ2, and FNQ5 derivatives showed potent concentration-depending inhibitory effect on nitric oxide production, with an IC₅₀ value lower than 2 µM concentration at which they did not have toxic or scavenging effects. FNQ5 was the most active and selective derivative.

Conclusions: this is the first paper concerning the anti-inflammatory potential of tested synthetic compounds. Our results indicated that FNQ5 might be considered as useful potential anti-inflammatory molecule to treat inflammatory diseases related with nitric oxide overproduction.

Key words: furanonaphthoquinones, synthesis, *Tabebuia*, antiinflammatory agents, nitric oxide (NO•).

RESUMEN

Objetivo: describir la síntesis de análogos de furanonaftoquinonas aisladas del género *Tabebuia* y su efecto inhibitor en la producción de óxido nítrico.

Métodos: se obtuvo una serie de seis derivados a través de reacciones de cicloadición y se caracterizaron los productos por métodos espectroscópicos. Se evaluó la actividad biológica por su efecto en la producción del mediador proinflamatorio en macrófagos RAW 264.7 activados con lipopolisacárido. Para asegurar que los compuestos no interfirieran con la viabilidad celular, se evaluó su efecto citotóxico empleando el ensayo de metiltetrazolio. Adicionalmente, se evaluó el efecto captador del radical *in vitro*.

Resultados: los derivados FNQ1, FNQ2 y FNQ5 demostraron potente efecto inhibitorio en la producción de óxido nítrico de manera concentración-dependiente, con un valor de CI_{50} menor que 2 μ M, concentración a la que no ejercieron efectos tóxicos o captadores de radicales. FNQ5 resultó el compuesto más activo y selectivo.

Conclusiones: este trabajo es el primero que evalúa el potencial antiinflamatorio de los compuestos sintetizados. Los resultados indican que FNQ5 puede ser considerada como una molécula de uso potencial para el tratamiento de enfermedades inflamatorias que cursen con sobreproducción de óxido nítrico.

Palabras clave: uranonaftoquinonas, síntesis, *Tabebuia*, agentes antiinflamatorios, óxido nítrico (NO•).

INTRODUCTION

Inflammation is a complex and delicate mechanism composed of cellular immunity and biochemical mediators with interrelated biological effects that occur as a response to injury, infection, and stress.^{1,2} However, excessive production of inflammatory mediators during chronic inflammation contributes to the pathogenesis and development of some diseases such as cardiovascular and bowel diseases, cancer, diabetes, arthritis, and neurodegenerative disorders that affect a significant part of the human population.^{3,4} Steroids and non-steroidal anti-inflammatory drugs (NSAID) are the most clinically important drugs to treat inflammatory diseases, which are associated with a high incidence of adverse effects.⁵ This justifies research directed at study and identification of new active substances safer and more effective to prevent and treat inflammatory disorders and related conditions.

Tabebuia spp. (Bignoniaceae) includes approximately 100 species, known as strictly woody, found in tropical rain forest areas throughout Central and South America.^{6,7} The products obtained from the inner bark of several species of this genus are popularly called Taheebo, Lapacho, Pau d'Arco, and Ipe roxo, and have been traditionally used as a poultice or concentrated tea for treating a variety of diseases associated with an inflammatory component.^{8,9} This traditional use has been validated by studies which identified several fractions obtained from these plants with anti-inflammatory, astringent, anti-bacterial and anti-fungal effect.^{8,10,11} In our previous report, we demonstrated the anti-inflammatory activity of ethanol extracts and some fractions from *Tabebuia rosea* and *Tabebuia ochracea* using the 12-*o*-tetradecanoylphorbol-13-acetate induced ear edema in mice.¹² The major active compounds identified in the bark extracts from *Tabebuia* spp. include naphthoquinones,

furanonaphthoquinones (FNQ), anthraquinones, benzoic acid and benzaldehyde derivatives, iridoids, coumarins, and flavonoids.^{11,13} Among these compounds, FNQ derivatives are an important group of molecules, widely distributed in nature and associated with a broad range of biological activities.¹⁴

The experimental evidence has demonstrated the anti-inflammatory potential of FNQ by inhibition of cytokine release,^{15,16} mast cell degranulation,¹⁷ and nitric oxide (NO•) and prostaglandin E2 (PGE2) production by suppression of NF- κ B.¹⁸

In previous work, our research group reported the isolation and identification of a series of FNQ with antimalarial and immunomodulatory activity from the stem inner bark of several species of *Tabebuia* genus.¹⁹⁻²¹ However, the isolation yield of compounds from *Tabebuia* extracts was very low and limited their use in assays of biological activity. This situation encourages us to continue our studies obtaining five synthetic analogues of FNQ isolated from *Tabebuia* genus and the evaluation of their potential NO• inhibitory activity. Structural variations were introduced in order to reflect changes in the biological activity of dihydrofuranonaphthoquinone derivatives with different type and number of substitutions at the 2 and 3 position of the dihydrofuran ring.

METHODS

2-hydroxy-1,4-naphthoquinone (lawsone), cerium (IV) ammonium nitrate (CAN), silica gel for column chromatography (Merck grade 9385, 230-400 mesh), Dulbecco's Modified Eagle Medium (DMEM), L-glutamine, antibiotics (Penicillin-Streptomycin), dimethyl sulfoxide (DMSO), trypan blue, lipopolysaccharides from *Escherichia coli* serotype 0127:B8 (LPS), (N-[[3-(aminomethyl)phenyl]methyl]-ethanimidamide dihydrochloride (1400W), sodium nitrite (NaNO₂), sodium nitroprusside (SNP), N-(1,1-naphthyl)ethylenediamine dihydrochloride, sulfanilamide, and Phosphate Buffer Saline tablets were purchased from Sigma-Aldrich (St. Louis, MO, USA). Fetal bovine serum (FBS) was obtained from GIBCO (Gaithersburg, MD, USA), and 3-(4,5-dimethyl-thiazol-2-yl)-2,5-diphenyl-tertzolium bromide (MTT) from Calbiochem® (USA). Organic solvents were analytical grade and obtained from Mallinckrodt Baker (San Diego, CA, USA).

SYNTHESIS OF FNQ COMPOUNDS

The analogues of FNQ isolated from *Tabebuia* genus, were synthesized according to the methods previously reported.²²⁻²⁴ The synthetic route used to prepare the different molecules employed 2-hydroxy-1,4-naphthoquinone (lawsone), as starting material, and was based on cycloaddition reactions [3+2] in the presence of cerium (IV) ammonium nitrate (CAN), alkenes, alkynes and acetonitrile as solvent (Figure 1). All reactions were conducted under inert atmosphere conditions and products purified by column chromatography on silica gel.

Compounds were unambiguously characterized by spectroscopy methods and uncorrected melting points (m.p.) determined by Differential Scanning Calorimetry-DSC7 (Perkin-Elmer, USA), assigning the corresponding structures and their physicochemical characteristics that in all cases coincided with those previously reported: 2-methyl-2-vinyl-2,3-dihydronaphtho[2,3-*b*]furan-4,9-dione (FNQ1), bright orange needles (27.3 %), m.p.: 149 °C; 2-methyl-2-vinyl-2,3-dihydronaphtho[1,2-*b*]furan-4,5-dione (FNQ2), yellow needles (35.8 %), m.p.: 2-phenylnaphtho[2,3-*b*]furan-4,9-dione (FNQ3), yellow crystals (47 %), m.p.: 158 °C; 2-phenyl-2,3-dihydronaphtho[1,2-*b*]furan-4,5-dione (FNQ4), dark orange crystals (27 %), m.p.:

115 °C; 2,2,3,3-tetramethyl-2,3-dihydronaphtho[2,3-*b*]furan-4,9-dione (FNQ5), yellow crystals (48.9 %) m.p.: 91 °C; 2,2,3,3-tetramethyl-2,3-dihydronaphtho[1,2-*b*]furan-4,5-dione (FNQ6), orange crystals (12.5 %) m.p.: 101 °C. FNQ5 and FNQ6 compounds could not be resolved into their corresponding stereoisomers.

All of the test compounds were dissolved in dimethylsulfoxide (DMSO) to obtain a stock solution, and stored as small aliquots at -20 °C. The compounds were diluted serially to the appropriate final concentration with supplemented culture medium, just before cell exposure with final concentration ranging from 10 µM to 0.01 µM, except for FNQ3 which was tested at a maximum concentration of 1 µM because its low solubility did not permit preparation of an appropriate stock solution. The final percentage of DMSO was adjusted to 0.1 % (v/v). FNQ6 was not tested because of its low yield.

DETERMINATION OF THE PARTITION COEFFICIENT (logP)

Lipid solubility was theoretically calculated for each of the molecules evaluated by their logP values, using a free online server available on the website <http://www.vcclab.org/lab/alogs/>.

CELL CULTURE

The murine RAW 264.7 macrophage-like cell line was obtained from the American Type Culture Collection (TIB-71; Rockville, MD, USA) and routinely cultured in DMEM supplemented with 2 mM L-glutamine, antibiotics (100 IU/mL of penicillin-100 µg/mL streptomycin) and heat-inactivated fetal bovine serum (FBS) at 37 °C in a humidified atmosphere containing 5 % CO₂ and 95 % air.

MTT REDUCTION ASSAY

Cytotoxic effect of test compounds was evaluated using the 3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide test (MTT assay).²⁵ RAW 264.7 macrophages (2 x 10⁵ cells/mL) were plated into 96-well plates and allowed to grow at 37 °C in 5 % CO₂ atmosphere. Subsequently, the culture medium was replaced with various concentrations of test compounds for 30 min, followed by stimulation with of *E. coli* lipopolysaccharide (LPS; 10 µg/mL) and incubated for 24 h. Triton X-100TM (2 %) was used as positive control. After incubation, treatment was retired and replaced by medium containing MTT (0.25 mg/mL). Four hours later, the medium was carefully aspirated and formazan crystals were dissolved in DMSO. Optical Density at 550 nm (OD₅₅₀) was measured using a microplate reader (Multiscan EX Thermo®). Percentages of cell survival relative to control group were calculated, as well as the concentration that reduces survival to 50 % (LC₅₀).

NO• PRODUCTION

The evaluation of the NO• inhibitory activity was performed in a manner similar to that described for cell viability. In brief, RAW 264.7 cells were seeded in 24-well plates (2 x 10⁵ cells/mL) and allowed to grow at 37 °C in 5 % CO₂ atmosphere. The adherent cells were treated for 30 min with various concentrations of test compounds or (N-[[3-(aminomethyl)phenyl]methyl]-ethanimidamide dihydrochloride (1 400W), as positive control, and stimulated with LPS (10 µg/mL). Twenty four hours later, culture supernatants were collected and stored at -20 °C until use.

Nitrite (NO_2^-) accumulation, as an indicator of $\text{NO}\bullet$ production in the medium, was determined by the Griess method.²⁶ Briefly, 100 μL of supernatants were mixed with an equal volume of Griess reagent (1:1 mixture of 0.1 % N-(1-naphthyl) ethylenediamine dihydrochloride and 1 % sulfanilamide in 5 % H_3PO_4) and incubated at room temperature for 5 min. The OD_{550} of the samples was measured using a microplate reader (Multiscan EX Thermo®). The amount of nitrite in the samples was calculated from a standard curve of sodium nitrite (NaNO_2). Percentage of inhibition was calculated against cells that were not treated but were induced with LPS. The concentration that inhibited the LPS-stimulated $\text{NO}\bullet$ production by 50 % (IC_{50}) was determined.

SCAVENGING OF $\text{NO}\bullet$

This assay was carried out in order to evaluate the effect of synthetic FNQ on $\text{NO}\bullet$ in a cell-free system. Sodium nitroprusside (SNP) was used to generate $\text{NO}\bullet$ in an aqueous solution at physiological pH, which was detected by Griess reagent after reaction with dissolved oxygen to form NO_2^- . Scavengers of $\text{NO}\bullet$ compete with oxygen, leading to reduce production of NO_2^- .²⁷ Test compounds were incubated with 1 mL of SNP (5 mM) in PBS at 25 °C for 120 min. Samples (100 μL) were incubated at room temperature for 5 minutes with an equal volume of Griess reagent. The OD_{550} of the samples was measured using a microplate reader (Multiscan EX Thermo®) and compared with standard solutions of NaNO_2 to calculate the nitrite concentration.

STATISTICAL ANALYSIS

All values are expressed as mean \pm standard error of the mean (S.E.M.). LC_{50} and IC_{50} values were calculated using non-linear regression analysis and expressed as mean and its 95 % confidence interval. Selectivity index (SI) was calculated as the ratio $\text{LC}_{50}/\text{IC}_{50}$. Data were analyzed using one-way analysis of variance (ANOVA), followed by Dunnett's post hoc test, to determine the differences between groups. Values of $p < 0.05$ were considered significant.

RESULTS

SYNTHESIS OF FNQ COMPOUNDS

Six analogues of FNQ isolated from *Tabebuia* genus were prepared as shown in Figure 1, obtaining a mixture of linear and angular conformational isomers, with yields varying between 12.5 % and 48.9 %. The lowest yields were obtained with compounds FNQ4 and FNQ6, 27 % and 12.5 %, respectively, which indicates the dominance of linear isomer. As previously reported, the high regioselectivity of cyclization can be easily explained by the mechanism of the reaction: initial formation of radicals followed by oxidation and formation of carbocations intermediates which react intramolecularly with the hydroxyl groups to produce linear and angular isomers, *ortho*- and *para*-quinones (Figure 2). All compounds were characterized and their physicochemical properties coincided with those previously reported in the literature.^{22,23}

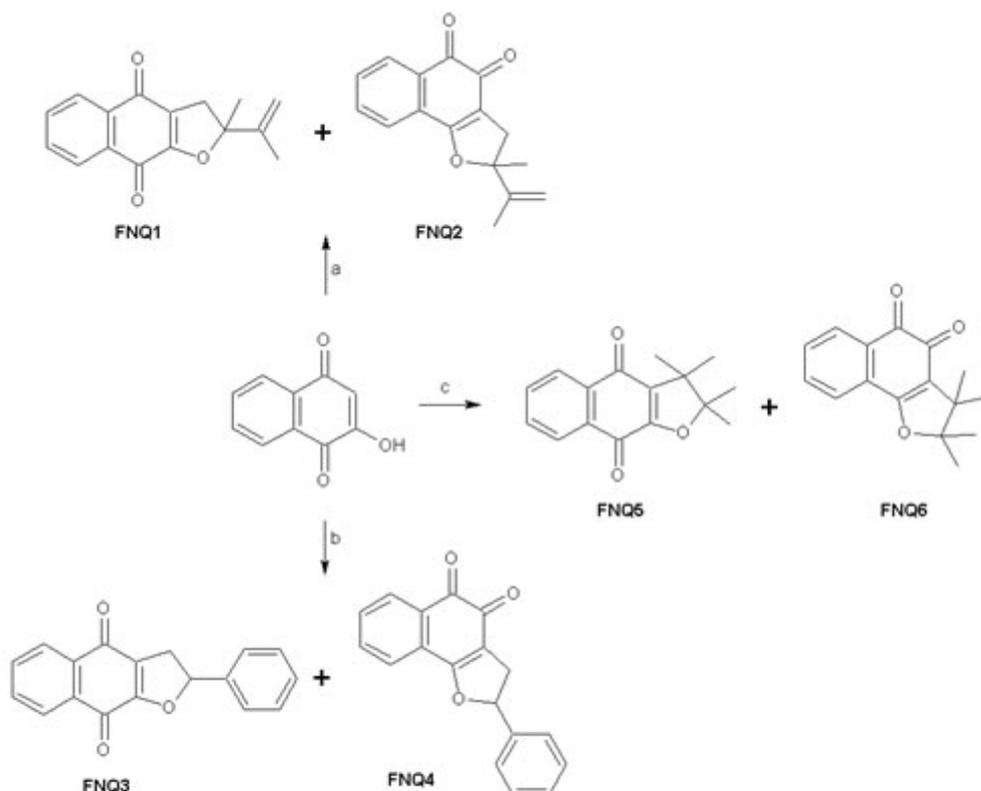


Fig. 1. Scheme of preparation of the synthetic analogues of FNQ isolated from the inner bark of species *Tabebuia* genus. Reagents and conditions: a) 2,3-dimethyl-1,2-butadiene, CAN, acetonitrile, 0°C; b) Styrene, CAN, acetonitrile, 0°C; c) 2,3-dimethyl-2-butene, CAN, acetonitrile, 0°C.

EFFECT OF SYNTHETIC ANALOGUES OF FNQ ISOLATED FROM *TABEBUIA* ON CELL VIABILITY

DMSO, used as vehicle, did not produce any significant alteration in cell viability, even at the highest concentration employed (0.1 %), which indicates that it did not interfere with the observed activity of compounds. Triton X-100™ (2 %), utilized as positive control, induced a mortality superior to 95%, confirming its utility in the assay. The MTT assay on RAW 264.7 macrophages showed that all of the tested FNQ produced reduction in cell viability proportional with increasing concentration of compounds, with no toxicity at 1 μM and percentages of cell viability below to 70 % at 10 μM, as can be seen in Figure 3.

Under our assay conditions, FNQ4 was the most toxic compound, LC₅₀ value of 1.99 (1.46-2.88) iM, followed by its linear conformer FNQ3, LC₅₀ of 2.18 (1.71-2.80) iM and compound FNQ5, LC₅₀ of 7.25 (8.45-5.80 μM). FNQ1 and FNQ2 showed some toxicity at 10 μM producing 36 % and 49.5 % of cell death, respectively. The test compounds were considered to be cytotoxic when the percentage of cell survival was less than 80 %. Therefore FNQ1 and FNQ2 were evaluated in the assays of anti-inflammatory activity at concentrations lower than 5 iM, and compounds FNQ3, FNQ4, and FNQ5 at concentrations lower than 1 μM.

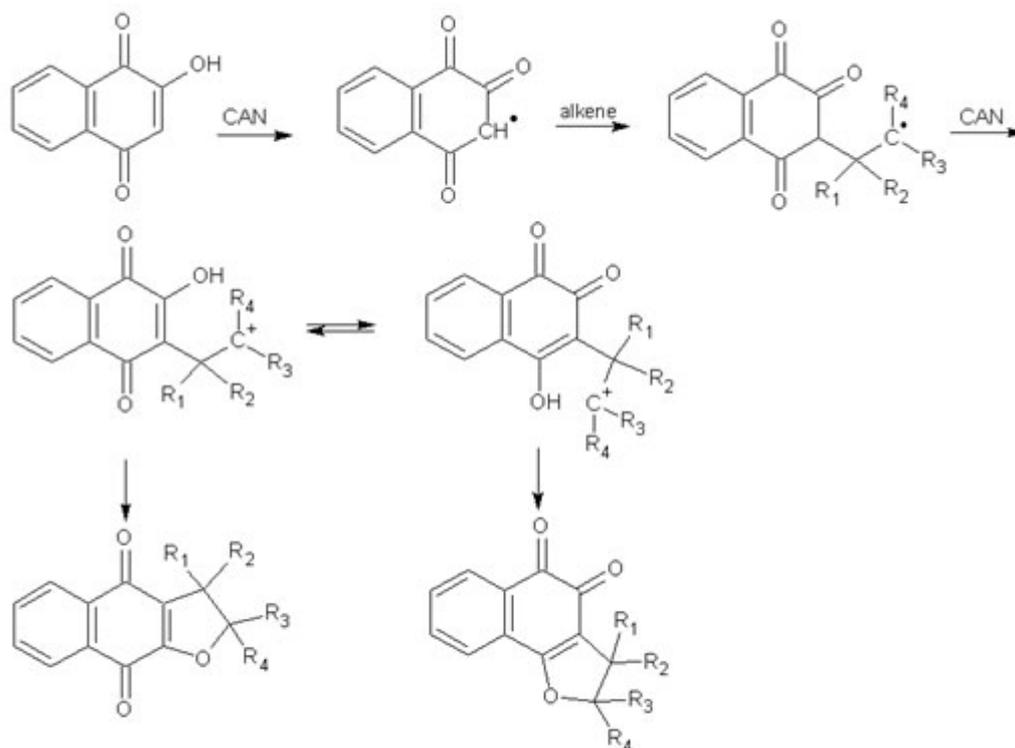


Fig. 2. Proposed mechanism for cycloaddition reaction catalyzed by CAN to obtain FNQ derivatives.

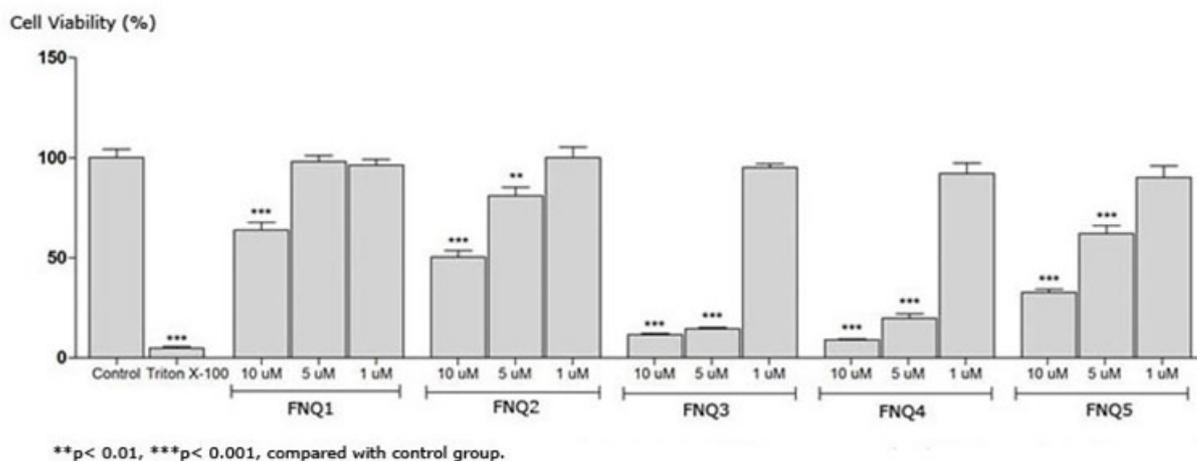


Fig. 3. Effect of synthetic analogues of FNQ isolated from *Tabebuia* species on cell viability of RAW 264.7 macrophages. Cells were incubated with various concentrations of test compounds for 24 h. Cell viability was measured with MTT assay. Each value represents mean \pm SEM ($n = 6$).

INHIBITORY EFFECT OF FNQ ON NO• PRODUCTION

The induction of RAW 264.7 macrophages into an inflammatory state by treatment with LPS caused synthesis and release of NO•. Nitrite was detected in the medium at a mean concentration of $28.72 \pm 7.18 \mu\text{M}$. Cells that were not treated with LPS released trace amounts of NO•. 1400W, a selective inhibitor of inducible nitric oxide synthase (iNOS), caused a decrease in LPS-induced NO production by more than 70 % at $10 \mu\text{M}$.

FNQ1, FNQ2 and FNQ5 derivatives inhibited significantly NO• production in a concentration-dependent manner, with an IC_{50} value of 1.84, 1.04 and $0.54 \mu\text{M}$, respectively, values lower than to the obtained for 1400W, 3.72 ($4.57\text{-}2.98 \mu\text{M}$). Results of anti-inflammatory effect of tested compounds are summarized in Table. In addition, the efficacy of inhibition of NO• production was estimated calculating a selectivity index (SI). FNQ1, FNQ2 and FNQ5, showed SI higher than 5, which indicates their potential as safe anti-inflammatory drugs.²⁸

Table. Effect of synthetic analogues of FNQ isolated from the inner bark of *Tabebuia* species, on cell viability and accumulation of NO• in the culture media of RAW 264.7 macrophages in response to LPS exposure

Compounds	Log P	$\text{LC}_{50} \pm$ confidence intervals (μM)	$\text{IC}_{50} \pm$ confidence intervals (μM)	SI
FNQ1	2.77	> 10	1.84 (1.57-2.19)	> 5.44
FNQ2	2.72	> 10	1.04 (0.94-1.20)	> 9.64
FNQ3	3.06	2.18 (1.71-2.80)	nd ^a	nd
FNQ4	3.84	1.99 (1.46-2.88)	nd	nd
FNQ5	2.94	7.25 (5.80-8.45)	0.54 (0.43-0.65)	13.77
1400W		>10	3.72 (2.98-4.57)	> 10.75

^a nd: not determined.

Each value represents mean \pm SEM (n= 6).

NO-SCAVENGING EFFECT

SNP releases large amounts of NO• at physiological pH in aqueous solution. As expected, caffeic acid showed a significant effect as a scavenger of NO• (59.58 %). Results in Figure 4 indicated that the co-incubation of SNP with tested compounds at $10 \mu\text{M}$ did not diminished the levels of nitrite in the medium, indicating that suppression of NO• release shown by FNQ1, FNQ2 and FNQ5 can be directly attributed to blocking NO• production in LPS-stimulated RAW 264.7 macrophages.

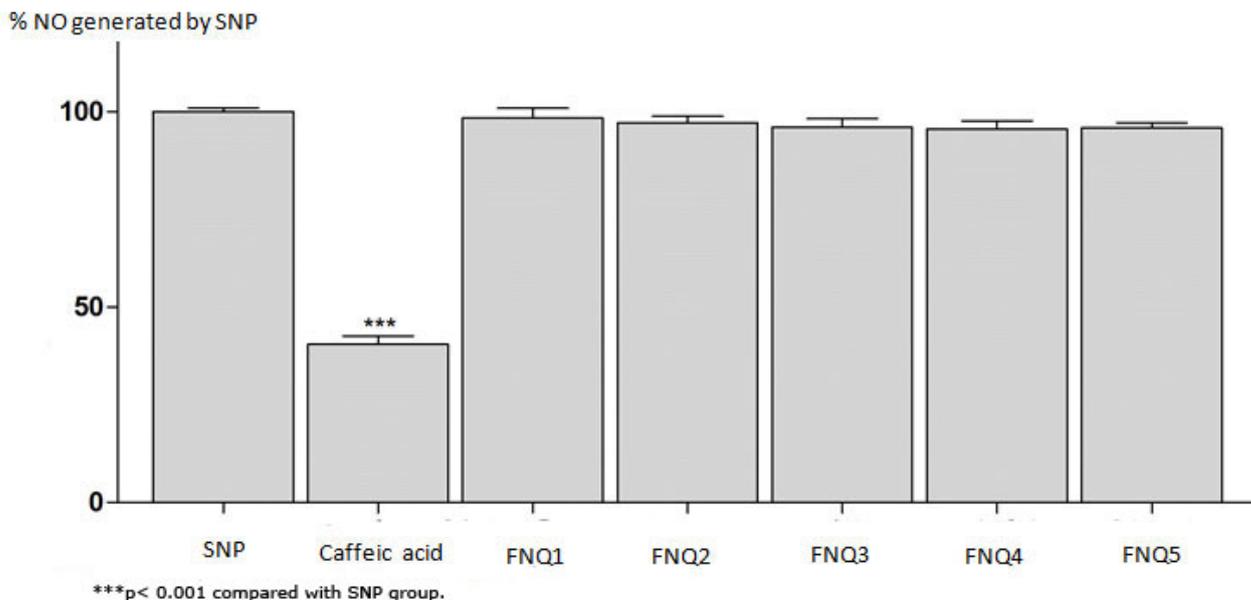


Fig. 4. Scavenging effect NO• generated from sodium nitroprusside (SNP) by synthetic analogues of FNQ isolated from *Tabebuia* (10 μM). Caffeic acid (50 μM) was employed as positive control.

DISCUSSION

Macrophages, named in this way because of their important role in phagocytosis of external intruders, antibody-antigen complexes and cellular debris, are fundamental in inflammatory and repair processes. They are considered the main immune effector cells, playing a pivotal role in the initiation, maintenance, and regulation of inflammation and innate immune response, the immediate arm of the immune system.^{29,30} During inflammation, macrophages are activated and produce reactive oxygen species (radical superoxide $O_2^{\bullet-}$, hydrogen peroxide H_2O_2 , the highly reactive hydroxyl radical $\bullet OH$, and $NO\bullet$), arachidonic acid metabolites, and lysosomal enzymes which are necessary to perform their phagocytic function.³¹

$NO\bullet$ is biosynthesis from the amino acid L-arginine in a process catalyzed by a family of oxidoreductases, called nitric oxide synthases (NOS). In NOS family, the inducible type (iNOS) is implicated in the pathophysiology of several chronic inflammatory conditions and can be activated in response to pro-inflammatory signals, such as cytokines and lipopolysaccharide (LPS) in various types of cells, including macrophages by binding TLR-4 receptor.^{32,33} $NO\bullet$ has multiple biological effects including vasodilatation and smooth muscle relaxation, inhibition of platelet aggregation, neuronal transmission, elimination of microorganisms and tumor cells, and it is recognized as an unique mediator of the inflammation and apoptosis. In small quantities, $NO\bullet$ acts protecting and repairing gastrointestinal mucosa, whereas abnormally high levels of this mediator has been described in a variety of pathological processes such as circulatory shock, carcinogenesis and chronic inflammation.³⁴ Furthermore, the experimental data support the idea that compounds inhibiting overproduction of $NO\bullet$ are potential anti-inflammatory agents with beneficial effects.³⁵ The purpose of this study was to evaluate the potential anti-inflammatory activity of synthetic analogues of FNQ isolated from the inner bark of species of the genus *Tabebuia* by determining $NO\bullet$ production by LPS-stimulated RAW 264.7 murine macrophages.

Quinones cyclization to FNQ derivatives, has allowed the preparation of compounds with a wide spectrum of biological activities. Therefore, this type of reactions is one of their most interesting characteristic, which has stimulated the development of several cyclization methods.³⁶⁻⁴⁰ Some of these reactions produced very low yields, situation that forced to find other alternatives. In our case, we employed reactions of cyclization of a variety of alkenes and alkynes in presence of lawsone. Reactions were mediated by CAN employing acetonitrile as solvent (Figure 1).⁴¹

The first precaution when studying the direct effect of a compound on NO• production is to guarantee that it does not cause cell death, decreasing the number of NO• producing cells. Thus, the effect on cell viability should be assessed.⁴² MTT assay revealed that all of the tested compounds produced toxic effect at 10 µM, being FNQ3 the most toxic one (Figure 3). The elevated toxicity of compounds was expected because, in general, tumor cells, as RAW 264.7 macrophages are sensitive to the effect of FNQ derivatives at low micromolar range.^{28,43} In addition, literature reports that compounds with FNQ moiety exhibits potent antiproliferative effect against several human neoplastic cell lines through three major mechanisms of toxicity: stimulation of oxidative stress, alkylation of cellular nucleophiles, and intercalation into DNA.⁴⁴⁻⁴⁷ Therefore, our findings suggest that FNQ4 is a promissory lead compound for the design, synthesis and development of new chemotherapeutics agents.

Whit regard to the NO• inhibitory activity of tested compounds, we found that FNQ1, FNQ2, and FNQ5 derivatives have a potent and selective inhibitory effect on NO• production in LPS-induced macrophages RAW 264.7 (Table), producing a mean IC₅₀ value less than to that obtained for the positive control drug 1 400W, a highly selective inhibitor of iNOS, which is 5000 times more selective to this isoform than endothelial NOS (eNOS), and 200 times more than neuronal NOS (nNOS).⁴⁸ This inhibitory effect might contribute significantly to the mechanism of anti-inflammatory activity reported for extracts and fractions obtained from species of *Tabebuia*.^{8,12,49} Also, this observation is in agreement with previous reports of anti-inflammatory activity of naphthoquinone derivatives which act inhibiting NO• production through down-regulation of iNOS expression by inactivating NF-κB.^{42,50,51} Based on previous experimental evidence, we strongly believe that NO• inhibitory activity of tested compounds is produced by blocking NF-κB activation, which is necessary for LPS induction of the iNOS promoter. The mechanisms by which FNQ derivatives could interfere with the activation of NF-κB remains to be elucidated.

Our results, positioned FNQ5 as the most active, CI₅₀ 0.54 (0.65-0.43 µM), and selective (SI= 13.77) compound. This observation might be related to its higher lipophilicity due to the presence of methyl substituents, which favors both the entrance of the compound in the cell and the establishment of hydrophobic bonds with a potential active site, leading to diverse biological effects, including selectivity among bioreceptors, increased potency, and protection against enzyme metabolism, etc.^{42,52} Thus, FNQ5 constitutes a promissory compound to investigate the *in vivo* effect and as a lead compound to design new derivatives which could suppress with greater potency and selectivity overproduction of NO•. On the other hand, we recommend further *in vivo* studies to establish the toxicity associated with the strong inhibition of NO• production of FNQ5, as well as other toxic side effect related to the naphthoquinone moiety.⁵³⁻⁵⁵

Upon comparing the structure-activity relationship of the tested compounds, it could generally be noted that modifications on furan ring substituents induce changes in the inhibition of NO• production and cytotoxic activity. Bulky aromatic substituents at 2-position of furan ring increased the cytotoxic activity (FNQ3 and FNQ4) and reduced the inhibition of NO• production. Whereas dimethyl groups (FNQ5) or vinyl groups (FNQ1 and FNQ2) at the same position, instead reduced the cytotoxicity (LC₅₀ value

higher than 7 μM) and led to a significant increase of inhibition of NO• activity (IC_{50} value lower than 1.8 μM). On the other hand, the small difference of FNQ1-FNQ2 and FNQ3-FNQ4 activities, suggest that linear (1,4-FNQ) or angular (1,2-FNQ) conformation, is not decisive for toxicity or NO• inhibitory activity (Figure 3 and Table).

CONCLUSION

In this work, we obtained six synthetic furanonaphthoquinones analogues of natural products isolated from *Tabebuia* genus, which exhibited a promissory profile. FNQ5 was the most active compound of the series, causing reduction of NO• production by LPS-stimulated RAW 264.7 murine macrophages, without cytotoxic or NO• radical scavengers effects, probably due to the presence of methyl substituents on the furan ring that rise the lipophilicity. The NO• inhibitory activity of the tested molecules varied with the type and nature of the substituents, which probably might affect their penetration and interaction with the potential target site. Furthermore, the slight difference in the activities of FNQ1 and FNQ2, suggests that linear or angular conformation is not determining for NO• inhibitory activity. To our knowledge, this is the first report showing the biological activity of these synthetic molecules and their potential either as an alternative therapy to treat inflammatory diseases related with NO• overproduction, or as lead compounds to design new anti-inflammatory drugs.

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